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stoichiometry stumper #1 So, Suzanne did not have enough benzene to create the amount of ethyl benzene she claimed to have made. But she did have enough benzene to create the same amount of di-chloro benzene that killed her husband..

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Solved: Stoichiometry Stumper #1 You Are A
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are a forensic scientist. You are
investigating a murder involving poison.

Stoichiometry Stumper #1

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additional experience and endowment by
spending more cash. nevertheless when?
complete you receive that you require to get
those all needs with having significantly
cash?

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Stoich_Stumpers - Stoichiometry Stumper#1 You are a ...

Name: _____ WML Held 12-03 Stoichiometry Stumper #1 You are a forensic scientist. You are investigating a murder involving poison. The victim was poisoned with a compound

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called di-chloro benzene whose formula is $C_6H_4Cl_2$. Autopsy results show that the victim's body contained about 22 g of the poison, but the actual amount could have been slightly higher due to tissue absorption.

5 - Stoichiometry Stumpers (pick 2 write 1) -
Name ...

\$1.25 Monday, February 17, 2014 Procedure: Is she lying? 1. Convert 15 grams into moles by dividing by its mass, 78.114 g C_6H_6 . 2. Multiply the moles by the mass of $C_6H_5CH_3$. 3. You will get an answer of 18 g. Math Three forensic scientists are investigating a

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murder involving

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... If 1 mole of amus is the same mass as 1 gram, and 1 hydrogen atom has a mass of 1 amu, then 1 mole of hydrogen atoms would have a mass of 1 $\hat{\text{€}}$...

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Sample Stoichiometry Problems And Answers
Honors Stoichiometry 5 Mass-to-Mass (3 Step)
Stoichiometry Problem Solving 1. Determine
the mass of lithium hydroxide produced when
.38g of lithium nitride reacts with water. $\text{Li}_3\text{N} + 3\text{H}_2\text{O} \rightarrow \text{NH}_3 + 3\text{LiOH}$ 2. Find the mass
of sugar ($\text{C}_6\text{H}_{12}\text{O}_6$) required to produce
1.82g of carbon dioxide gas. $\text{C}_6\text{H}_{12}\text{O}_6 + 2\text{C}$

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$2 \text{H}_2\text{O} + 2\text{CO}_2$.

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