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Solution From Coulomb ' s law of electrostatic attraction we know that the PE between two charges Q1 and Q2 separated by a distance r is given by  $Q_1 Q_2 / (4 \pi \epsilon_0 r^2)$  or First we consider the interaction between Na+ ion and 6Cl - ions at distance r Applying Coulomb ' s law we have  $Q_1 Q_2 / (4 \pi \epsilon_0 r^2) = 6e(12e) / (4 \pi \epsilon_0 r^2)$  or Similarly, we now consider 12 Na+ ions as second nearest neighbors at a distance r  $Q_1 Q_2 / (4 \pi \epsilon_0 r^2) = 12e(12e) / (4 \pi \epsilon_0 r^2)$  or and Cl ...

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Ernest Rutherford (1871-1937) 1.37 fSolutions to Principles of Electronic Materials and Devices: 2nd Edition (Summer 2001) Chapter 2 Second Edition ( 2001 McGraw-Hill) Chapter 2.2.1 Electrical conduction Na is a monovalent metal (BCC) with a density of 0.9712 g cm-3. Its atomic mass is 22.99 g mol-1.

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